List 3 ways of monitoring rates of a reaction	Sketch a graph to show how the amount of product made in a reaction changes over time	C5 Monitoring and Controlling Chemical Reactions	Links ← →
		C5.1 Controlling Reactions	 1.1 The Particle Model 3.1 Introduction to Chemical Reactions 5.2 Equilibria 6.1 Improving Processes and Products
Label on your sketch where the reaction has stopped happening		Explain what collision theory is	Describe how the rate of a reaction changes when the temperature is increased
How could you use this graph to calculate the rate of reaction?	Describe how the rate of a reaction changes when smaller lumps of a solid reactant are used		Use collision theory to explain why this happens
What is activation energy?	Use collision theory to explain why this happens	Describe how the rate of a reaction when a catalyst is added Use collision theory to explain why the happens	Describe how the rate of a reaction changes when concentration or pressure is increased