AQA Chemistry

| Question number | Answer | Marks | Guidance |
|--------------------|--|-------|---|
| 1 (a) (i) | 2 or two or second or [E] ² | 1 | |
| 1 (a) (ii) | 1 or one or first or [F] ¹ or [F] | 1 | |
| 1 (b) (i) | $k = 8.6 \times 10^{-4} / ((3.8 \times 10^{-2})^2 \times (2.6 \times 10^{-2}))$ | 1 | mark is for insertion of numbers into a correctly rearranged rate equ, $k =$ etc. AE (-1) for copying numbers wrongly or swapping two numbers |
| | = 22.9 (Allow 22.9 – 24 after correct rounding) | 1 | |
| | $mol^{-2} dm^6 s^{-1}$ | 1 | Any order. |
| 1 (b) (ii) | 6.8(2) × 10 ⁻³ (mol dm ⁻³ s ⁻¹) OR if their <i>k</i> is wrong, award the mark consequentially a quick check can be achieved by using their answer / their $k = 2.9768 \times 10^{-4}$ | 1 | Allow 6.8 × 10^{-3} to 6.9 × 10^{-3} |
| | Allow 2.9 – 3.1 × 10 for the mark | | Ignore units |
| 2 (a) | Experiment 2 4.5×10^{-4} Experiment 3 4.5×10^{-3} | 1 | Minimum 2 s.f. If three wrong answers, check their value of <i>k</i> in 2(b). |
| | Experiment 4 0.043 OR 4.3 ×10 ⁻² OR 0.044 OR 4.4 ×10 ⁻² | 1 | They can score all 3 marks if they have used their (incorrect) value of <i>k</i> . See below: Experiment 2 rate = $k \times (1.0125 \times 10^{-4})$ Experiment 3 [Q] = 0.02/ <i>k</i> Experiment 4 [P] = 0.0913/ \sqrt{k} |
| 2 (b) | $k = 5.0 \times 10^{-5} / ((2.5 \times 10^{-2})^2 \times (1.8 \times 10^{-2}))$ | 1 | Mark is for insertion of numbers into a correctly rearranged rate equation. If upside down, score only units mark from their <i>k</i> AE (-1) for copying numbers wrongly or swapping two numbers |
| | = 4.4(4) (allow 40/9) | 1 | |
| | $mol^{-2} dm^6 s^{-1}$ | 1 | Any order If <i>k</i> calculation wrong, allow units conseq to their <i>k</i> expression |
| 3 (a) | 2 or two or second | 1 | |

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| 3 (b) | $k = 1.24 \times 10^{-4} / (4.40 \times 0.82)$ | 1 | mark is for insertion of numbers into a correctly rearranged rate equ, $k = \text{etc}$ if upside down, (or use of I_2 data) score |
|-------|---|---|--|
| | $= 3.44 \times 10^{-5}$ (min 3 s.f.) | 1 | only units mark |
| | $mol^{-1} dm^3 s^{-1}$ | 1 | any order |
| 3 (c) | no change or no effect or stays the same or 1.24×10^{-4} | 1 | |
| 3 (d) | 1 or 2 or 1 and 2 | 1 | if wrong no further mark but mark on from no answer |
| | rate equation doesn't involve I_2 or only step which includes 2 species in rate equation | 1 | |
| 3 (e) | $H \xrightarrow{H} C \xrightarrow{+} C \xrightarrow{+} CH_3 \longrightarrow H \xrightarrow{H} C \xrightarrow{-} CH_3 + H^+$ | 1 | any second arrow loses the mark |